

# Per- and Polyfluoroalkyl Substances: History, Occurrence, and Future Treatment Prospects

Jessica L. Bennett, Manda-Tchonlla, Graham A. Gagnon



waterstudies.

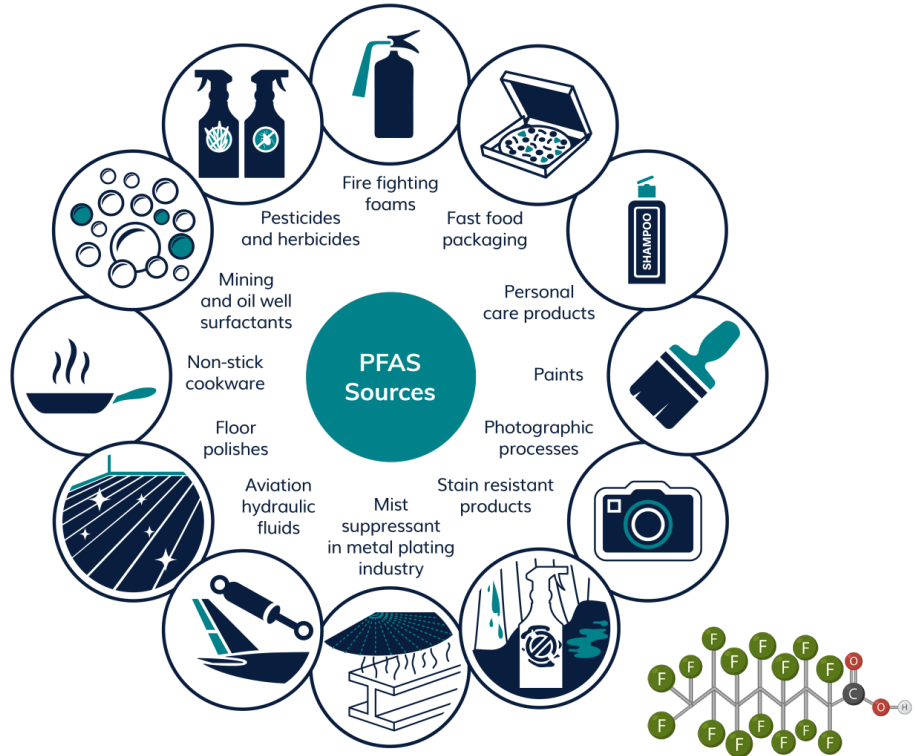
CENTRE FOR WATER RESOURCES STUDIES | DALHOUSIE UNIVERSITY

Faculty of Engineering  
Dalhousie University  
Halifax, NS Canada

March 26<sup>th</sup>, 2025

# Per- and Polyfluoroalkyl Substances (PFAS)

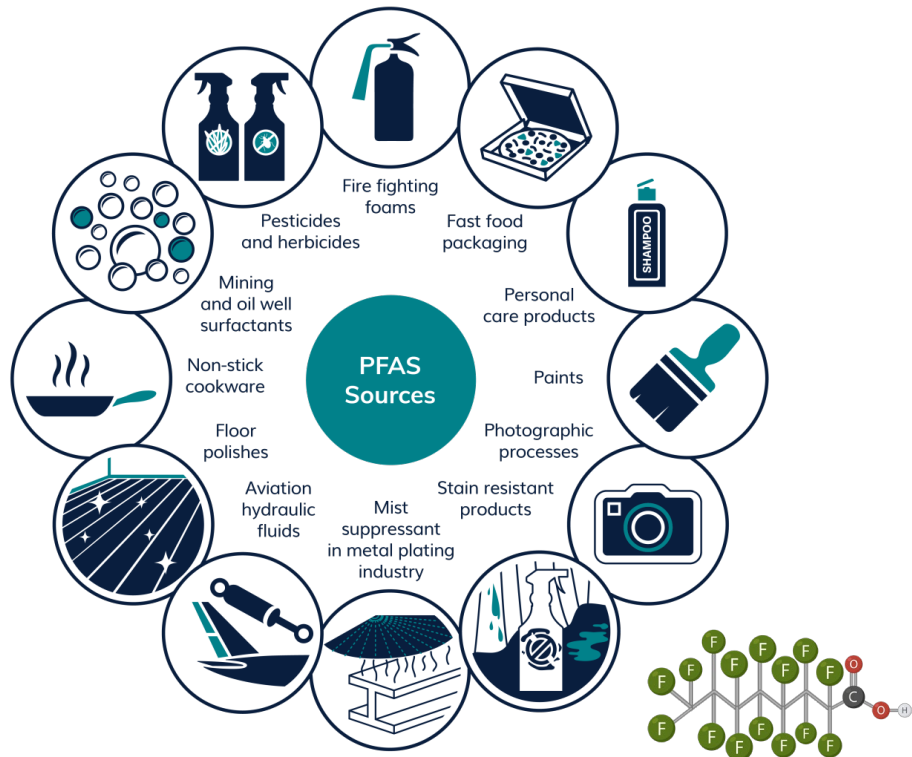
Group of > 14 000 synthetic organofluorine compounds



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Group of > 14 000 synthetic organofluorine compounds

Chemical properties have resulted in their use across consumer industries

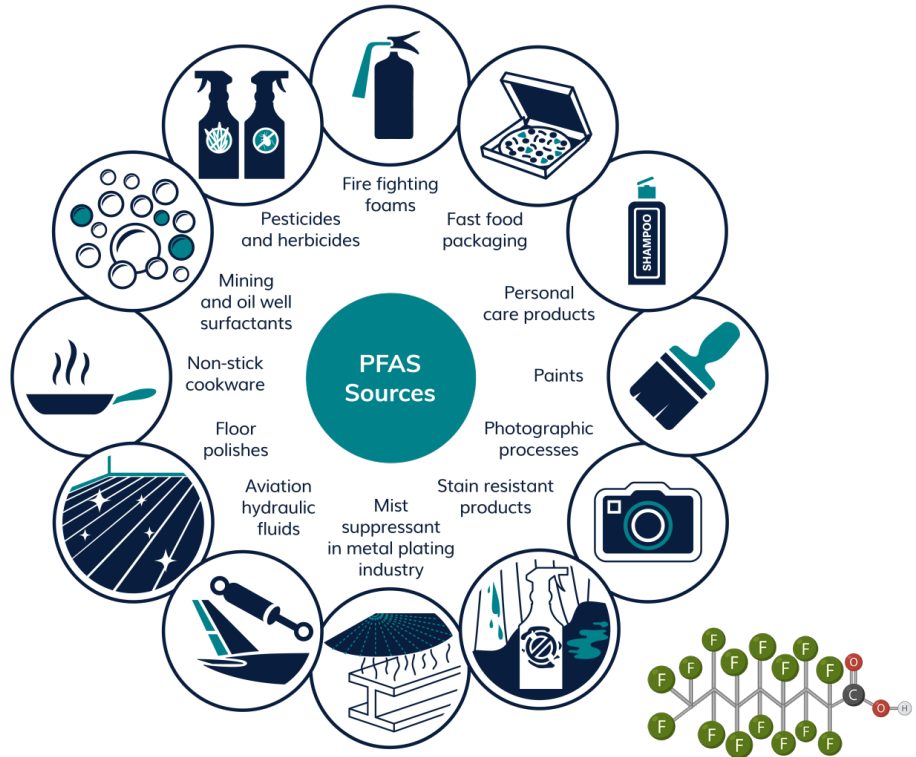


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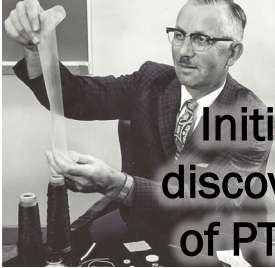
Chemical properties have resulted in their use across consumer industries

Inherent stability has resulted in environmental ubiquity



# History of PFAS

Roy Plunkett, 1938



Initial  
discovery  
of PTFE

1930s

1950s

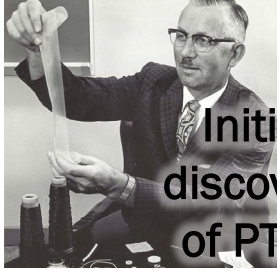
1970s

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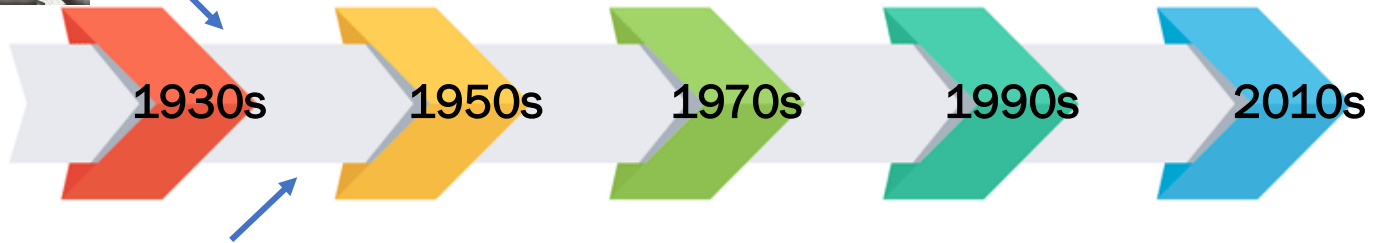
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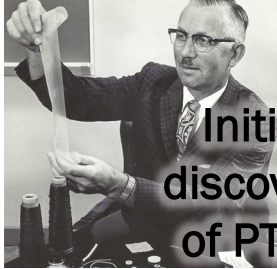


Use in Manhattan  
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1942 - 1947

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**3M**

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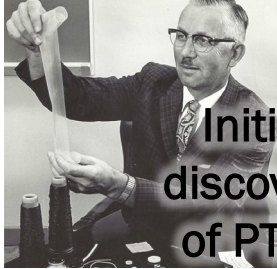


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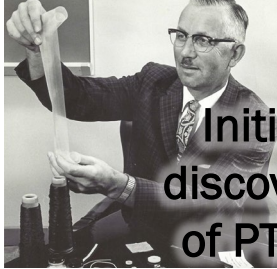


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Global Distribution of Perfluorooctane Sulfonate in Wildlife

JOHN P. GIESY\* AND KURUNTHACHALAM KANNAN  
Department of Zoology, National Food Safety and Toxicology Center, Institute for Environmental Toxicology, Michigan 48824

First study outlining widespread distribution of PFOS in environment  
2001

Development of AFFFs



# 3M

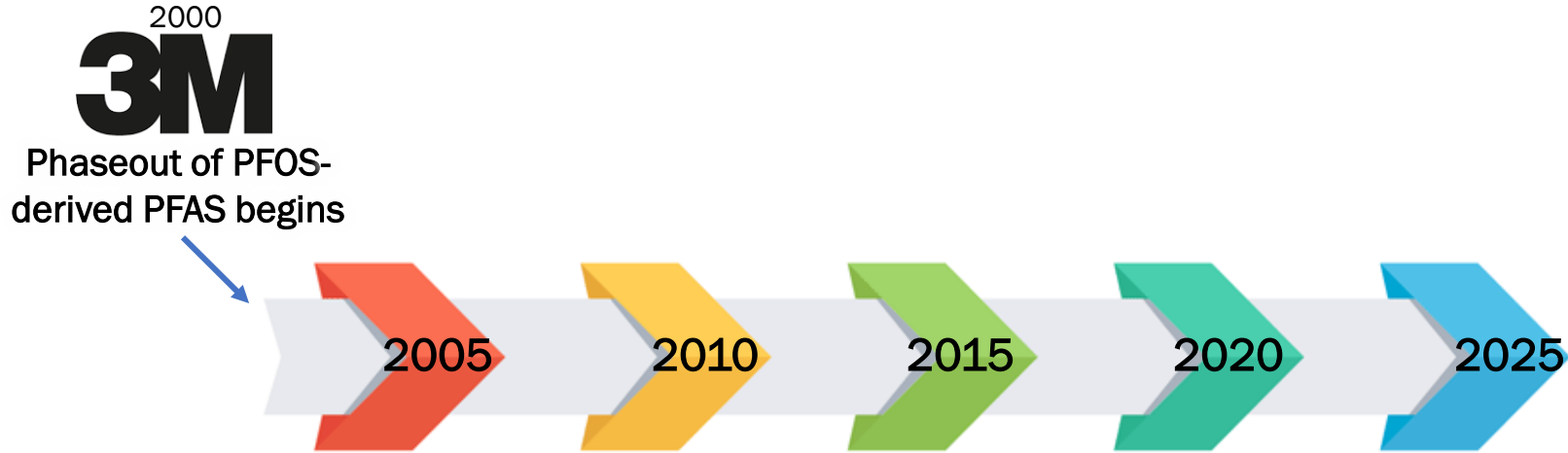
1963

Use in Manhattan Project

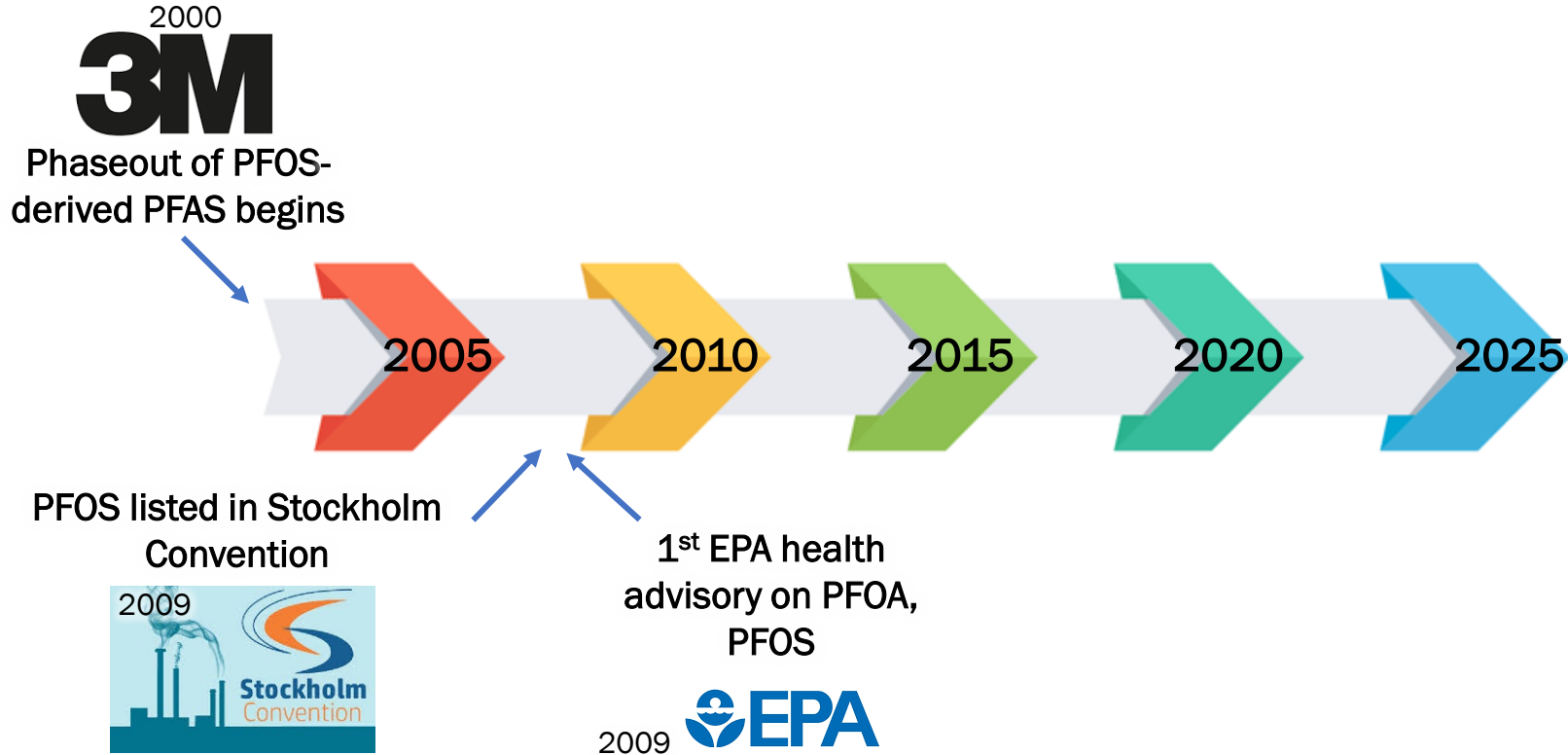


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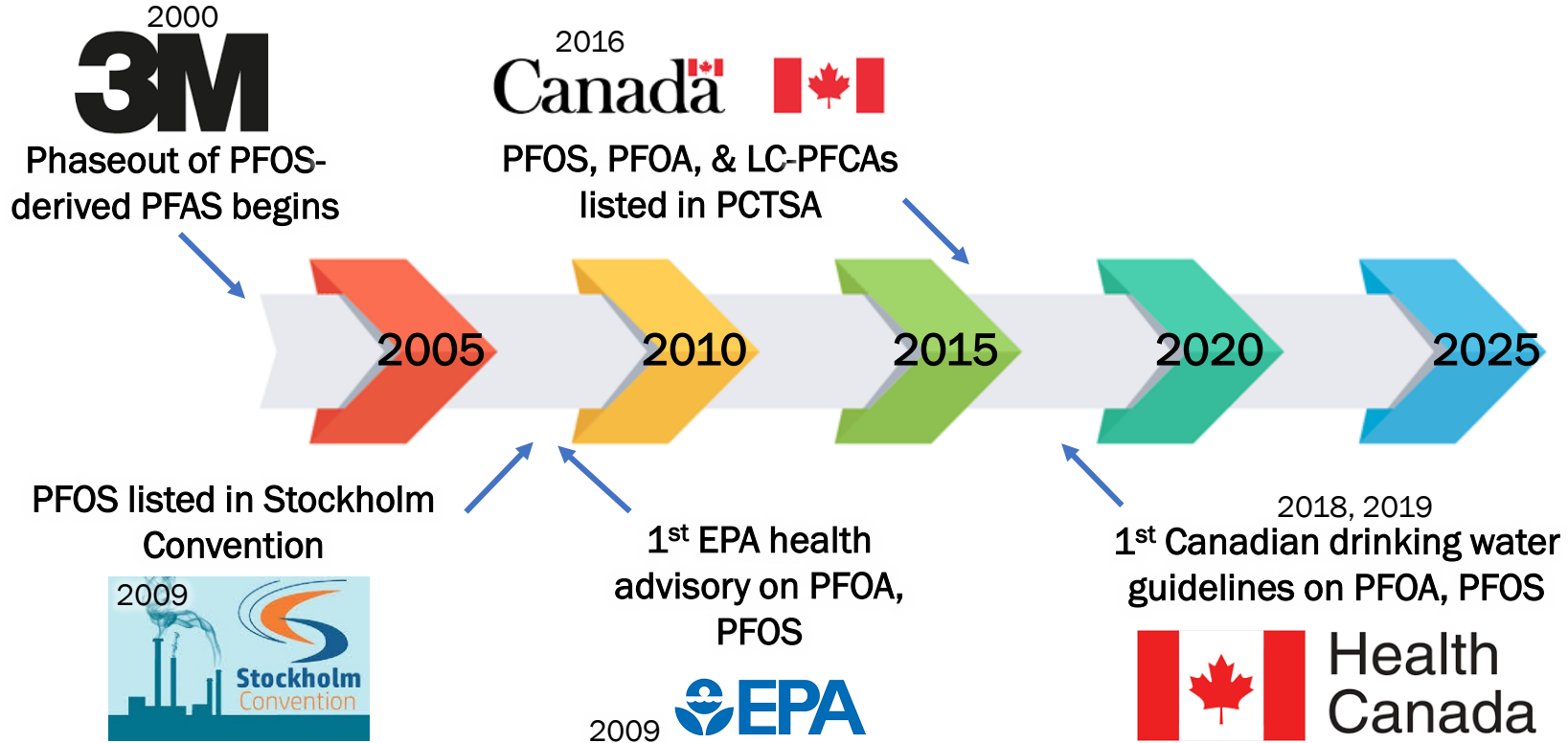
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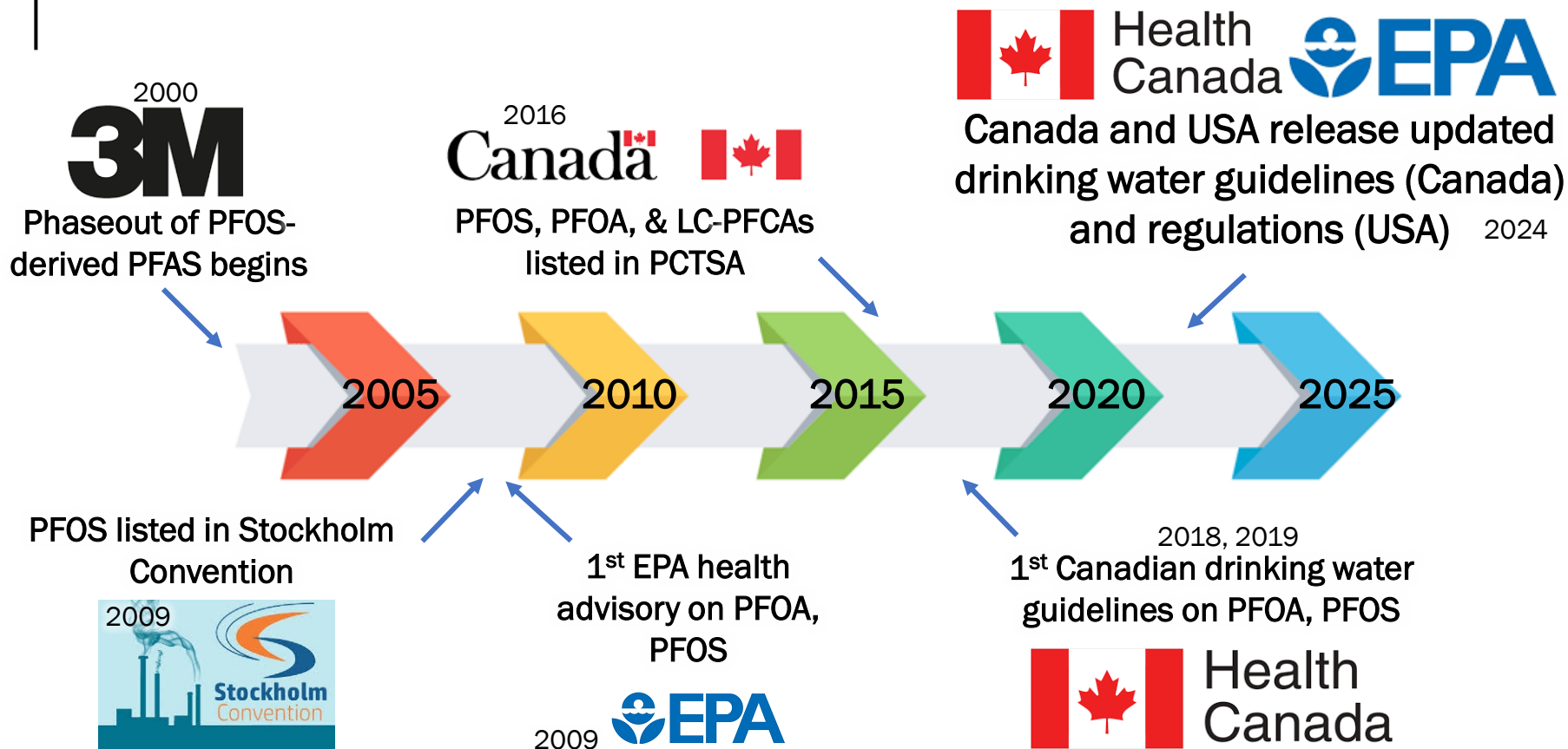
# History of PFAS



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# Current Guidelines and Regulations

Canadian Drinking Water Guidelines (August 2024)	$\Sigma$ 25 PFAS (as per EPA 533 or 1633)	$\Sigma$ 30 ng/L
Canadian Drinking Water Guidelines (Previous)	PFOS	600 ng/L
	PFOA	200 ng/L

US EPA National Primary Drinking Water Regulation	PFOA	4 ng/L
	PFOS	4 ng/L
	HFPO-DA	10 ng/L
	PFNA	10 ng/L
	PFHxS	10 ng/L
	PFAS Mixtures	Hazard Index of 1 (unitless)

Canada & USA have recently released stringent guidelines and regulations on PFAS in drinking water.

The background of the slide is a photograph of three wind turbines situated on a forested hill. The hill and turbines are reflected in a calm body of water in the foreground. The entire image has a blue color overlay. A yellow rectangular box with rounded corners is positioned in the center, containing the title text.

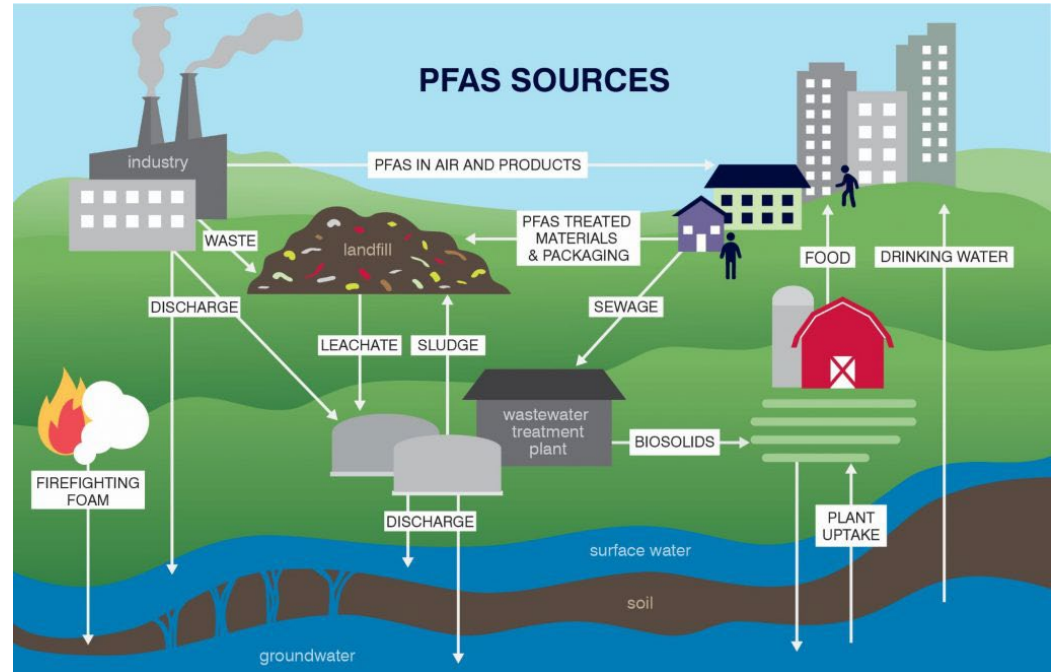
# **PFAS Occurrence in Canadian Waters**

# Transport of PFAS into the Environment

PFAS are released into the environment through anthropogenic sources

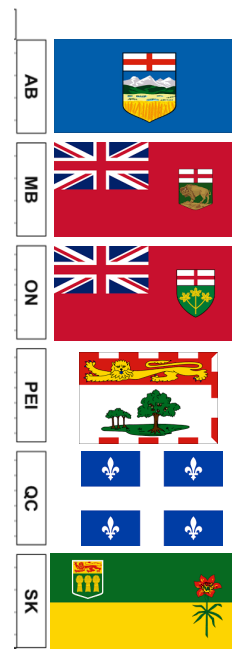
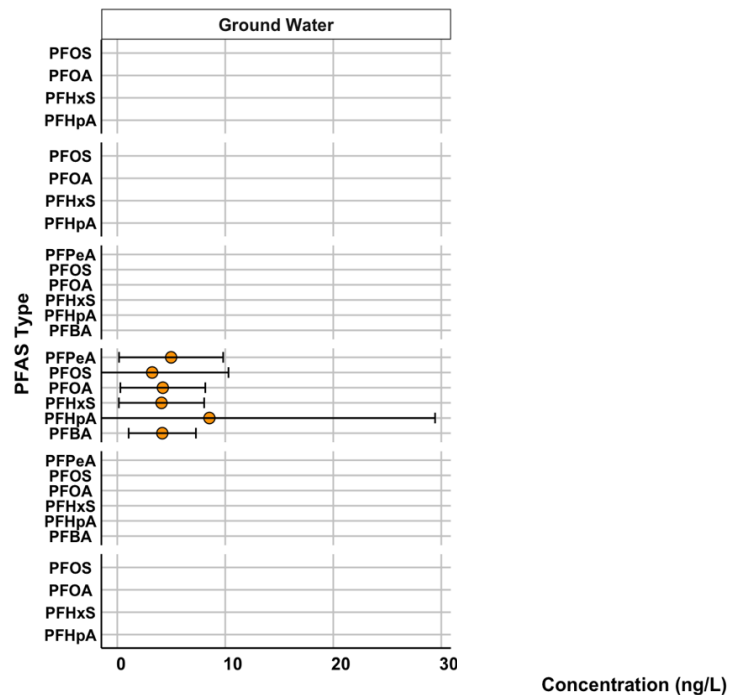
More highly industrious and populated areas are at higher risk of contamination

Transport to soils, surface water and groundwater can affect humans and wildlife



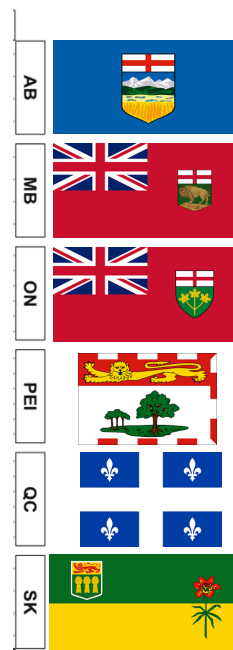
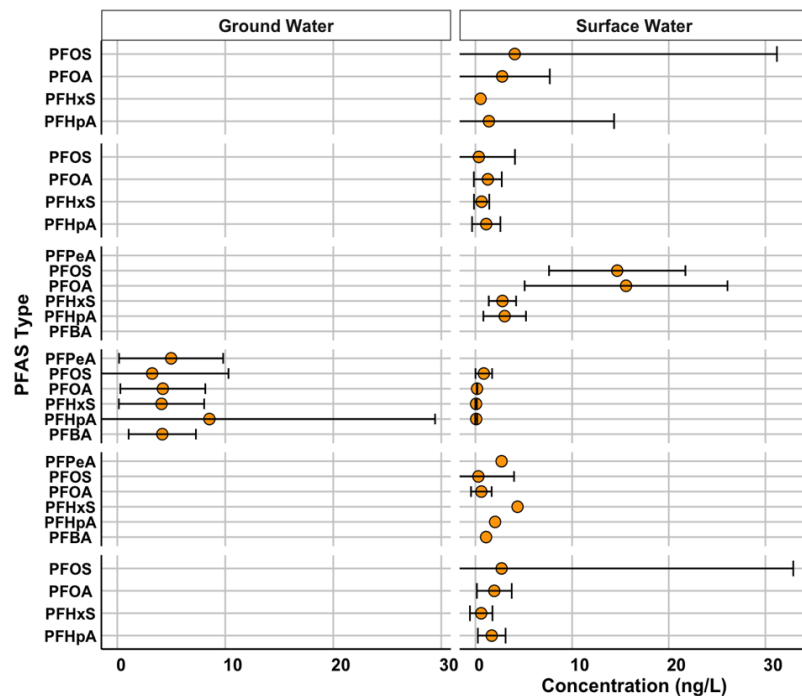
Weston & Sampson 2022

# Occurrence of PFAS in Canadian Waters



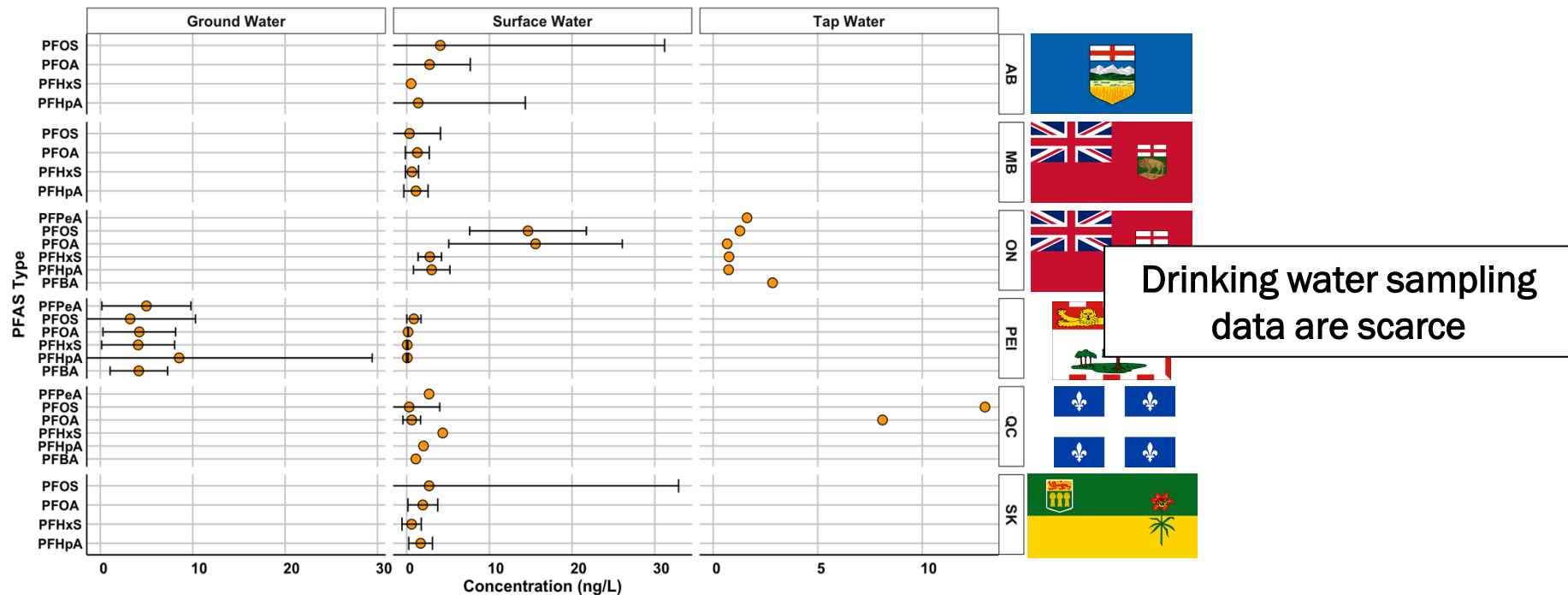
*Occurrence of PFAS in tap water, surface water, and groundwater in six Canadian Provinces*

# Occurrence of PFAS in Canadian Waters



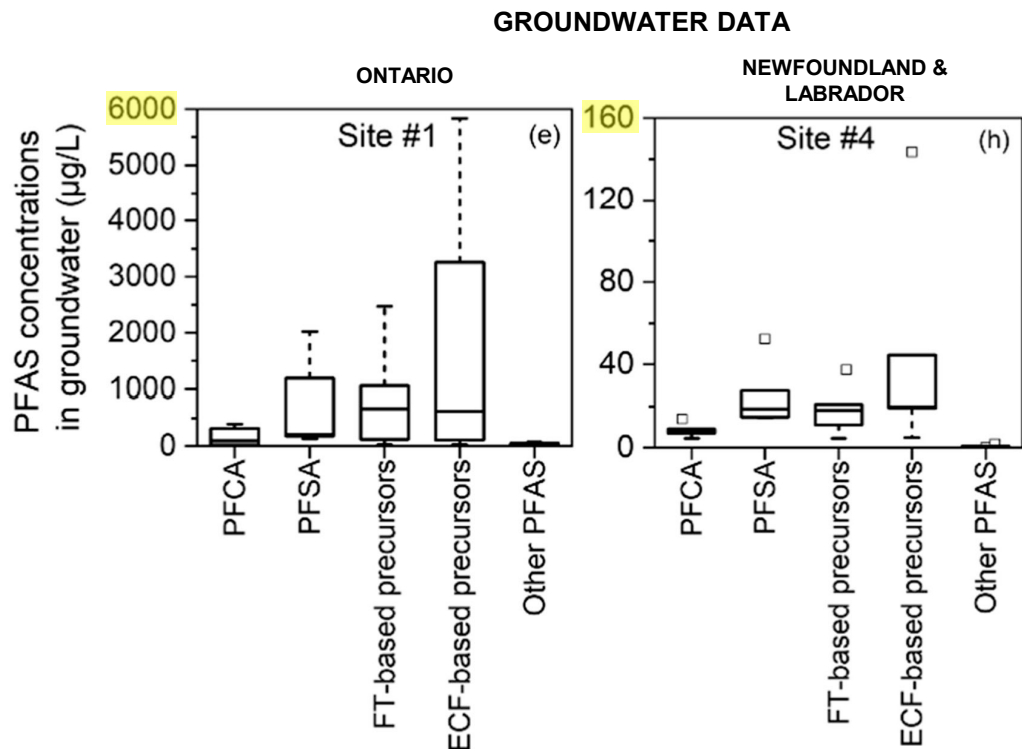
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# Occurrence of PFAS in Canadian Waters



Occurrence of PFAS in tap water, surface water, and groundwater in six Canadian Provinces

# Occurrence of PFAS in AFFF-Impacted Areas in Canada



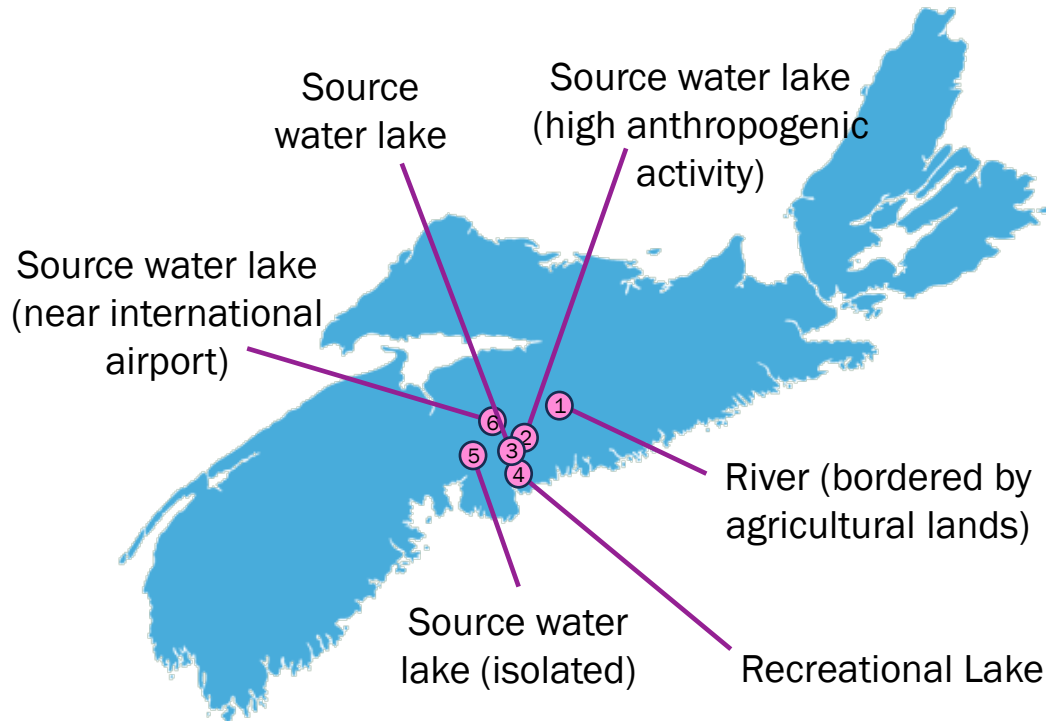
Historical use of AFFFs can result in substantial (**mg/L**) PFAS contamination

Groundwater sources may be particularly vulnerable

The background of the slide is a blue-tinted photograph. It shows a row of three wind turbines on a grassy hill. In the foreground, there is a calm body of water that reflects the sky and the turbines. The overall scene is serene and suggests a focus on environmental or renewable energy topics.

# **PFAS Research at Dalhousie University**

# Development of a PFAS Sampling Program in Nova Scotia



Occurrence of PFAS in Nova Scotian freshwater sources is not well understood

**Objective:** Determine PFAS occurrence in local freshwater systems to inform effects of human activity and land use patterns

# Development of a PFAS Sampling Program in Nova Scotia – Sampling Considerations

## UNACCEPTABLE MATERIALS



PTFE, GORTEX, waterproof materials



Personal care products, insect repellent, sunscreens containing PFAS



Glass, low-density polyethylene, aluminum foil, Teflon tape

Potential for unintentional contamination of samples is high

Care must be taken to avoid problematic materials and self contamination

Appropriate inclusion of sample blanks is important

# Development of a PFAS Sampling Program in Nova Scotia – Sampling Considerations

Blank Type	Rationale
Field Reagent Blank	Assesses PFAS contamination occurs during sample collection and handling
Equipment Rinse Blank	Verifies decontamination effectiveness of sampling equipment
Trip Blank	Detects background contamination from cooler/transportation recipient.

**Follow proper sample handling**

Use verified PFAS-free containers, adhere to preservation and shipping requirements, and ensure compliance with holding times.

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**Implement proper QA/QC measures**

Incorporate necessary blanks and standardize sampling techniques to ensure sample integrity

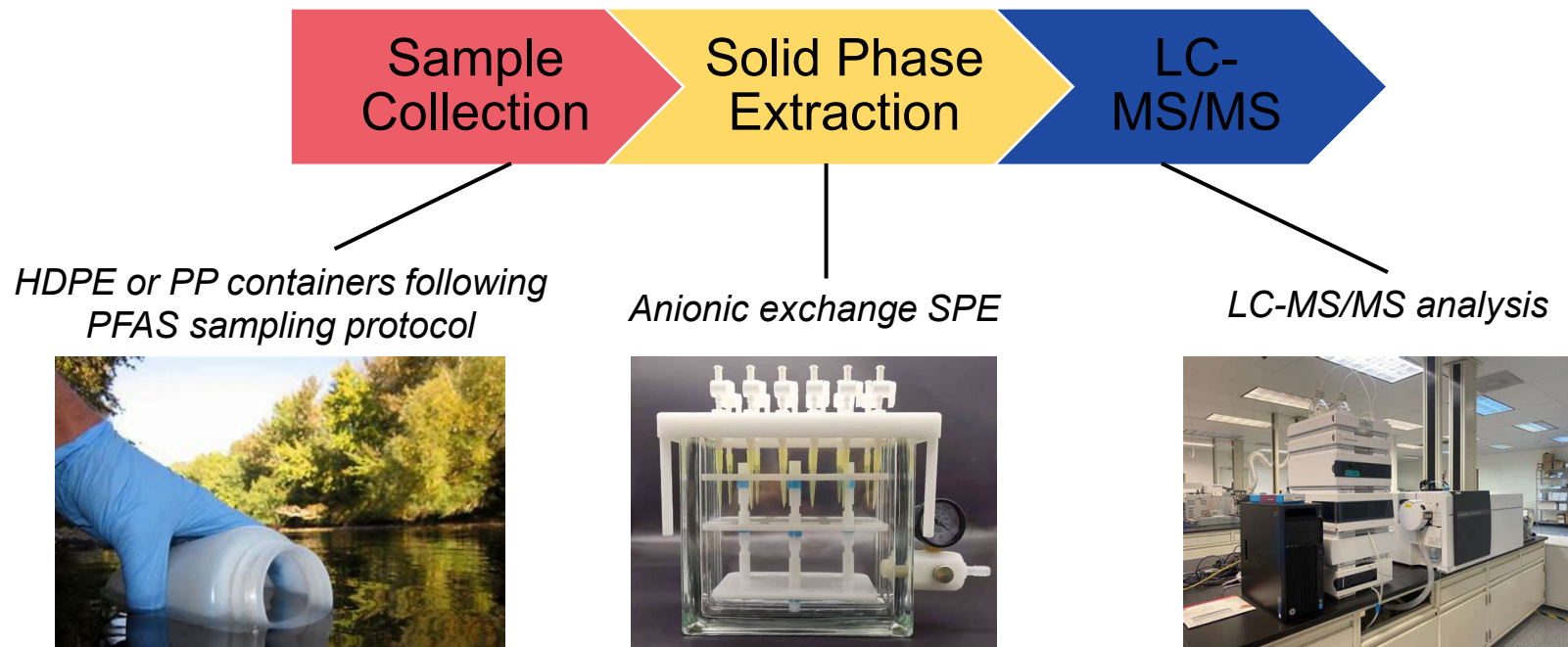
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**Maintain thorough documentation**

Record all sampling details, decontamination steps, deviations from protocols, and environmental conditions to support sampling effort

# EPA 533 for Analysis of PFAS in Drinking Water



Method reporting limits 0.5 – 1 ng/L for most PFAS

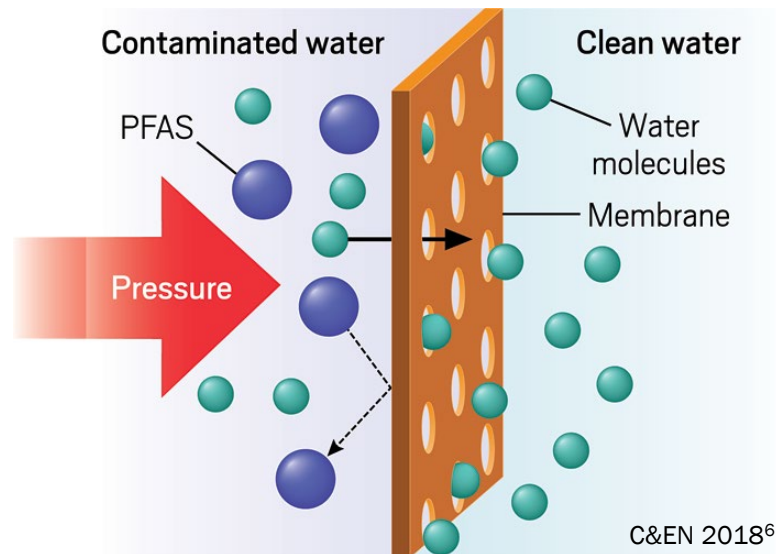
# PFAS Treatment Methods: Current State-of-the-Art

Separation methods (e.g., granular activated carbon, anion exchange, high-pressure membranes)

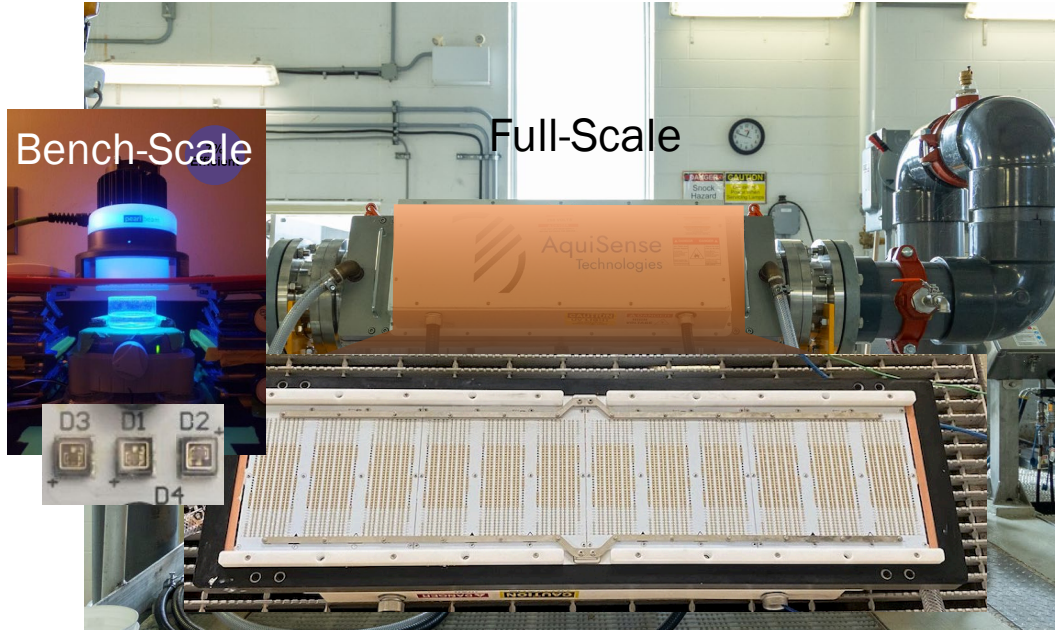
- Resulting PFAS-laden adsorbent or concentrate

Some pilot-scale evaluation of electrochemical technologies

- Have not been tested at scale

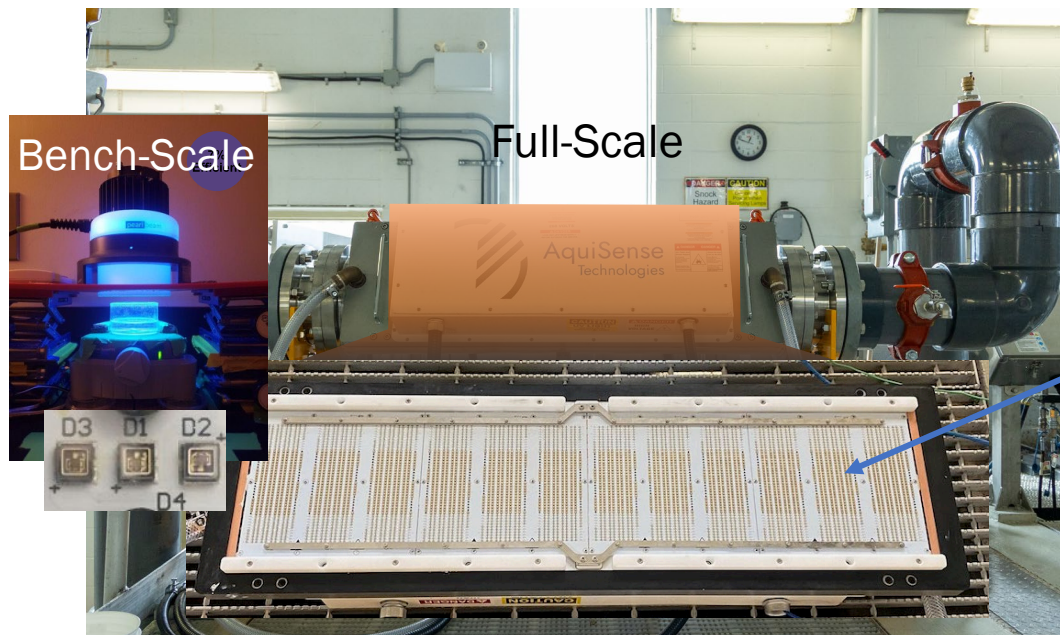


# Bespoke Treatment of Trace Organics using UV LEDs



UV LEDs have proven effective for disinfection of water at scale<sup>5</sup>

# Bespoke Treatment of Trace Organics using UV LEDs



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Light emission can be tailored to their use case

UV LEDs can serve a dual purpose for bespoke destruction of trace contaminants

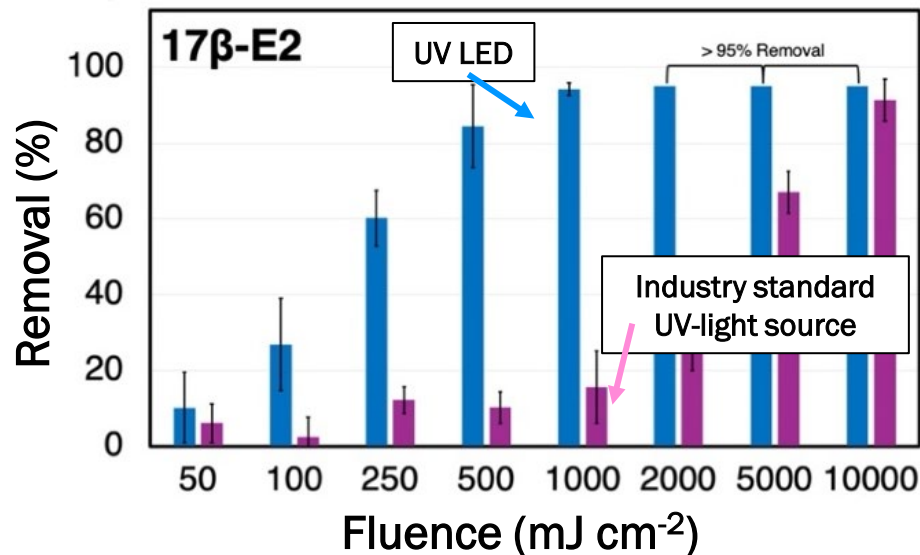
# Bespoke Treatment of Trace Organics using UV LEDs

UV LEDs substantially reduce  
fluence requirements for  
estrogen degradation

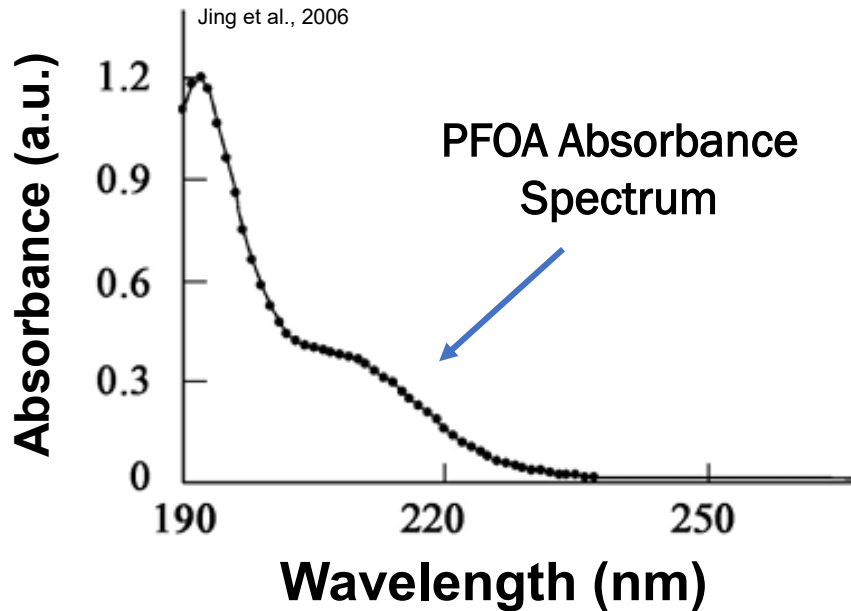
Energy requirements were  
reduced by > **four-fold**

275 nm = **2.66 kwh m<sup>3</sup> order<sup>-1</sup>**

MP UV = 11.31 kwh m<sup>3</sup> order<sup>-1</sup>



# Future Potential for Bespoke Treatment of PFAS using UV LEDs



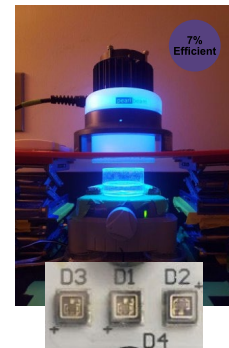
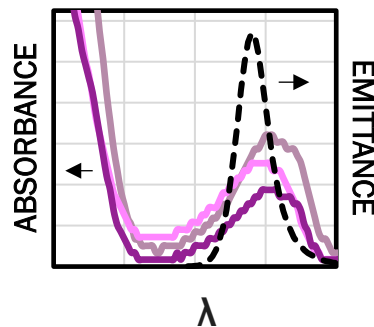
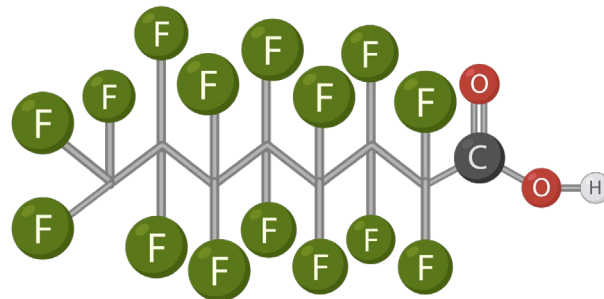
All PFAS absorb light in the far UVC region

Use of far-UVC LEDs could see bespoke remediation of legacy compounds

# Conclusions

Increased monitoring effort is needed to understand the occurrence of PFAS in Canadian waters

UV LEDs are a versatile technology that could see the development of chemical-free destruction of legacy contaminants



The research team acknowledges funding through our NSERC Alliance Partnerships.

