

Real-Time Water Quality Deployment Report

Rattling Brook Network

March 8, 2019 to April 17, 2019



Government of Newfoundland & Labrador
Department of Municipal Affairs and Environment
Water Resources Management Division
St. John's, NL, A1B 4J6 Canada

General

- Department of Municipal Affairs and Environment staff monitor the real-time web pages consistently.
- The reporting interval for Rattling Brook Big Pond in this report is a continuation of the deployment that began in December 2018. For convenience and comparative purposes, only the portion concurrent with Bridge and Plant Discharge deployment periods is considered in this report.
- During maintenance work on March 7th at Rattling Brook below Plant Discharge station, it was determined that water levels were far too low for a successful redeployment on March 8th. It was decided to wait for a melting event to reduce the level of ice cover and raise water levels. Normally, additional cable would be extended into the river channel, but this was not an option due to the snow cover. Deployment proceeded on March 15th.
- Hydrometric data included in this report is provisional and used only for illustrative purposes. Corrected and finalized data may be retrieved from the Water Survey of Canada website (<http://www.ec.gc.ca/rhc-wsc/>)*.

Maintenance and Calibration of Instrument

- As part of the Quality Assurance and Quality Control protocol (QAQC), an assessment of the reliability of data recorded by an instrument is made at the beginning and end of the deployment period. The procedure is based on the approach used by the United States Geological Survey.
 - Upon deployment, a QA/QC Sonde is temporarily deployed *in situ*, adjacent to the Field Sonde. Depending on the degree of difference between each parameter from the Field and QAQC sondes a qualitative rank is assigned (See Table 1). The possible ranks, from most to least desirable, are: Excellent, Good, Fair, Marginal, and Poor. A grab sample is also taken for additional confirmation of conditions at deployment and to allow for future modelling studies.
 - At the end of a deployment period, a freshly cleaned and calibrated QAQC Sonde is placed *in situ*, adjacent to the Field Sonde. Values are compared between all parameters and differences are ranked for placement in Table 1.

Table 1: Qualitative QAQC Ranking

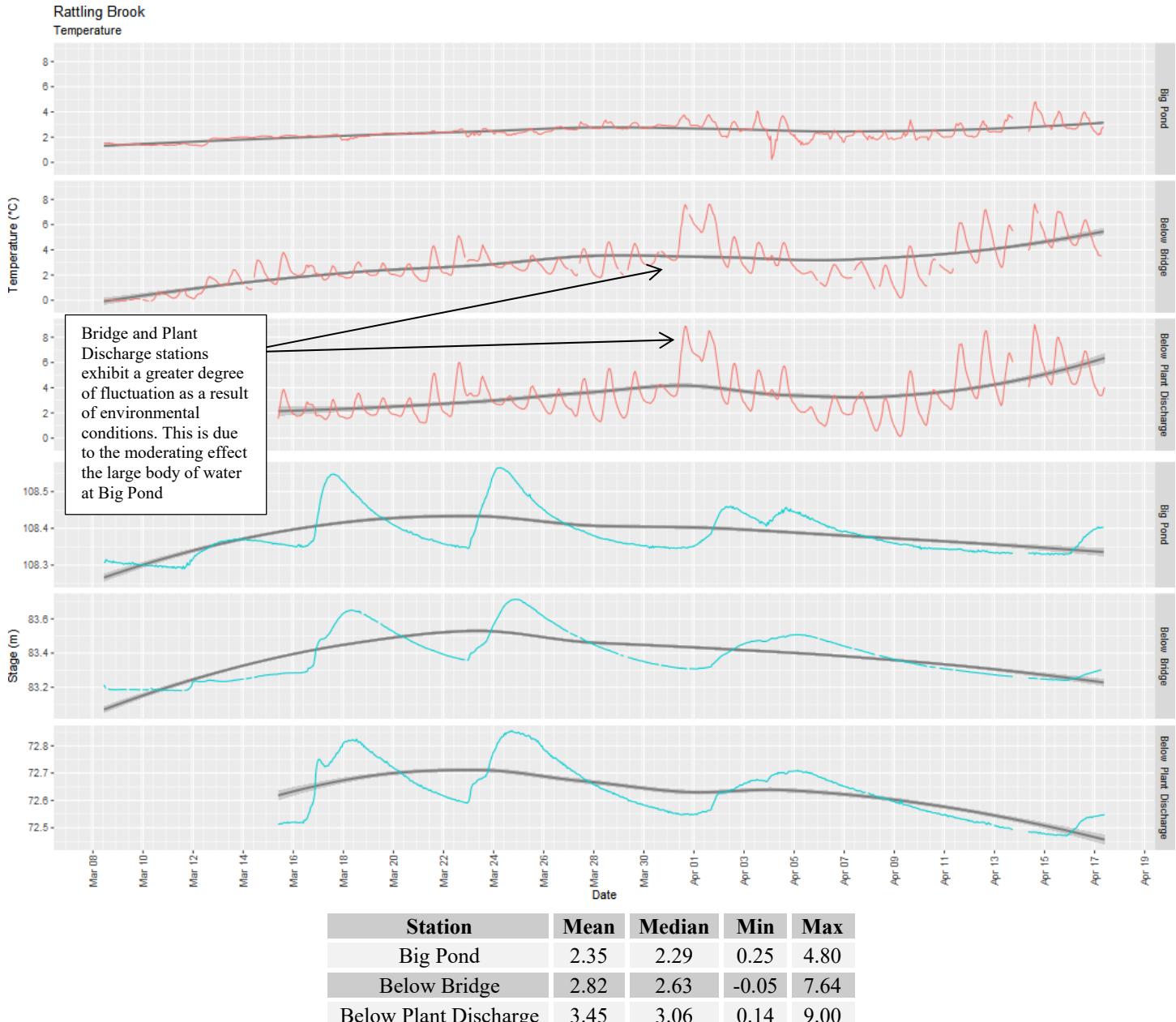
| Station | Date | Action | Comparison Ranking | | | | |
|---|------------|------------|--------------------|----------|--------------|------------------|-----------|
| | | | Temperature | pH | Conductivity | Dissolved Oxygen | Turbidity |
| Rattling Brook Big Pond | 2019-03-08 | Continued | NA | NA | NA | NA | NA |
| | 2019-04-17 | Removal | Marginal | Good | Good | Excellent | Excellent |
| Rattling Brook below Bridge | 2019-03-08 | Deployment | Excellent | Good | Fair | Excellent | Excellent |
| | 2019-04-17 | Removal | Excellent | Marginal | Good | Excellent | Excellent |
| Rattling Brook below Plant Discharge | 2019-03-15 | Deployment | Excellent | Good | Good | Excellent | Excellent |
| | 2019-04-17 | Removal | Excellent | Poor | Good | Excellent | Excellent |

- Since the deployment at Rattling Brook Big Pond began in December 2018, no QAQC rankings could be calculated for March 8th.
- During removal a “Poor” ranking was calculated at Discharge station; possibly a result of calibration drift.

Data Interpretation

Temperature

Water Temperature is a major factor used to describe water quality. Temperature has major implications on both the ecology and chemistry of a water body, governing processes such as the metabolic rate of aquatic plants and animals and the degree of dissolved oxygen saturation.



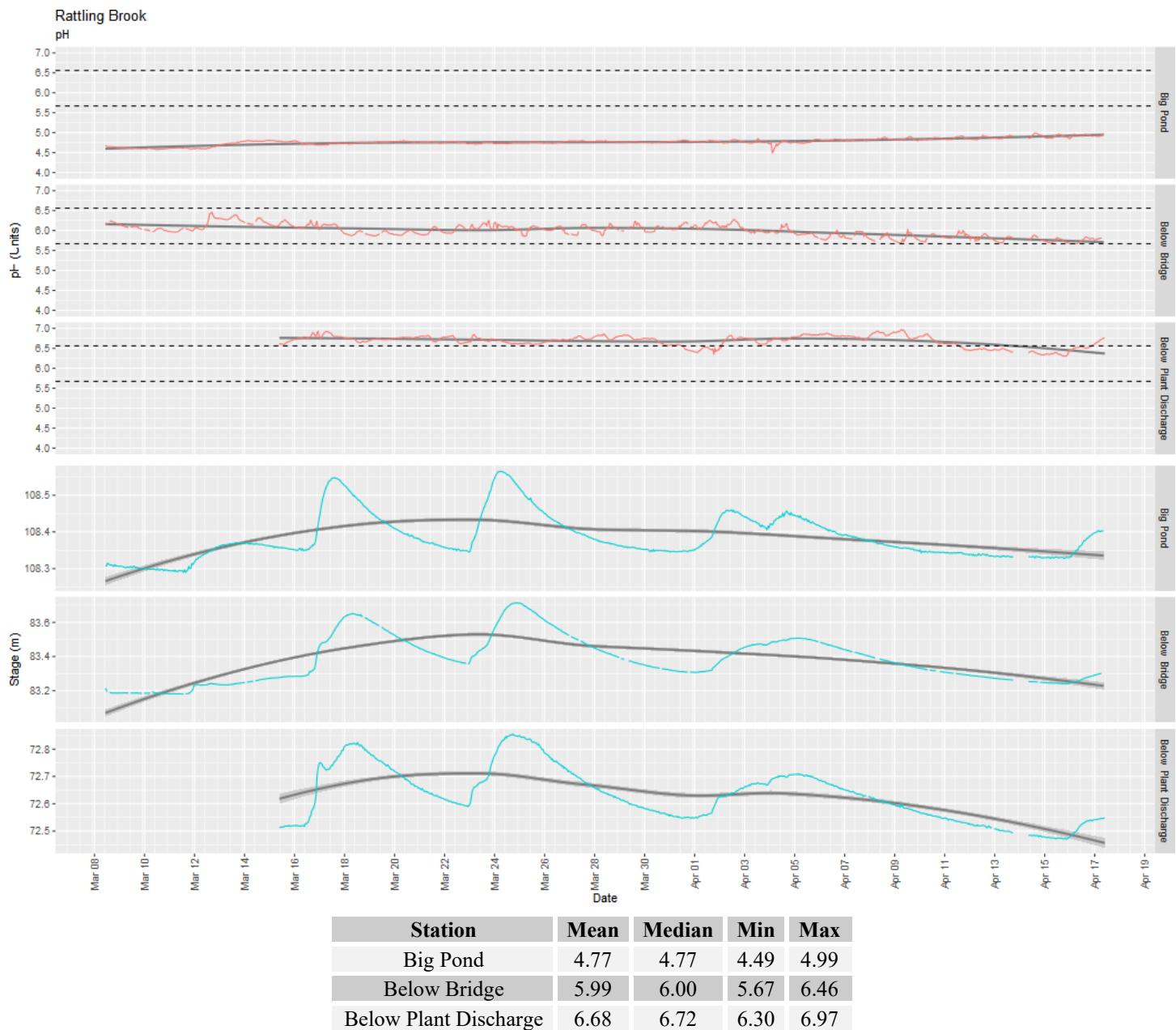
- Water temperatures at all stations indicated an upward trend as spring progressed. Bridge and Plant Discharge stations exhibited a resumption of diurnal cycling to a much larger degree than Big Pond station. This is likely due to the depth of cold water at Big Pond station.

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pH

pH is used to give an indication of the acidity or basicity of a solution. A pH of 7 denotes a neutral solution while lower values are acidic and higher values are basic. Technically, the pH of a solution indicates the availability of protons to react with molecules dissolved in water. Such reactions can affect how molecules function chemically and metabolically.

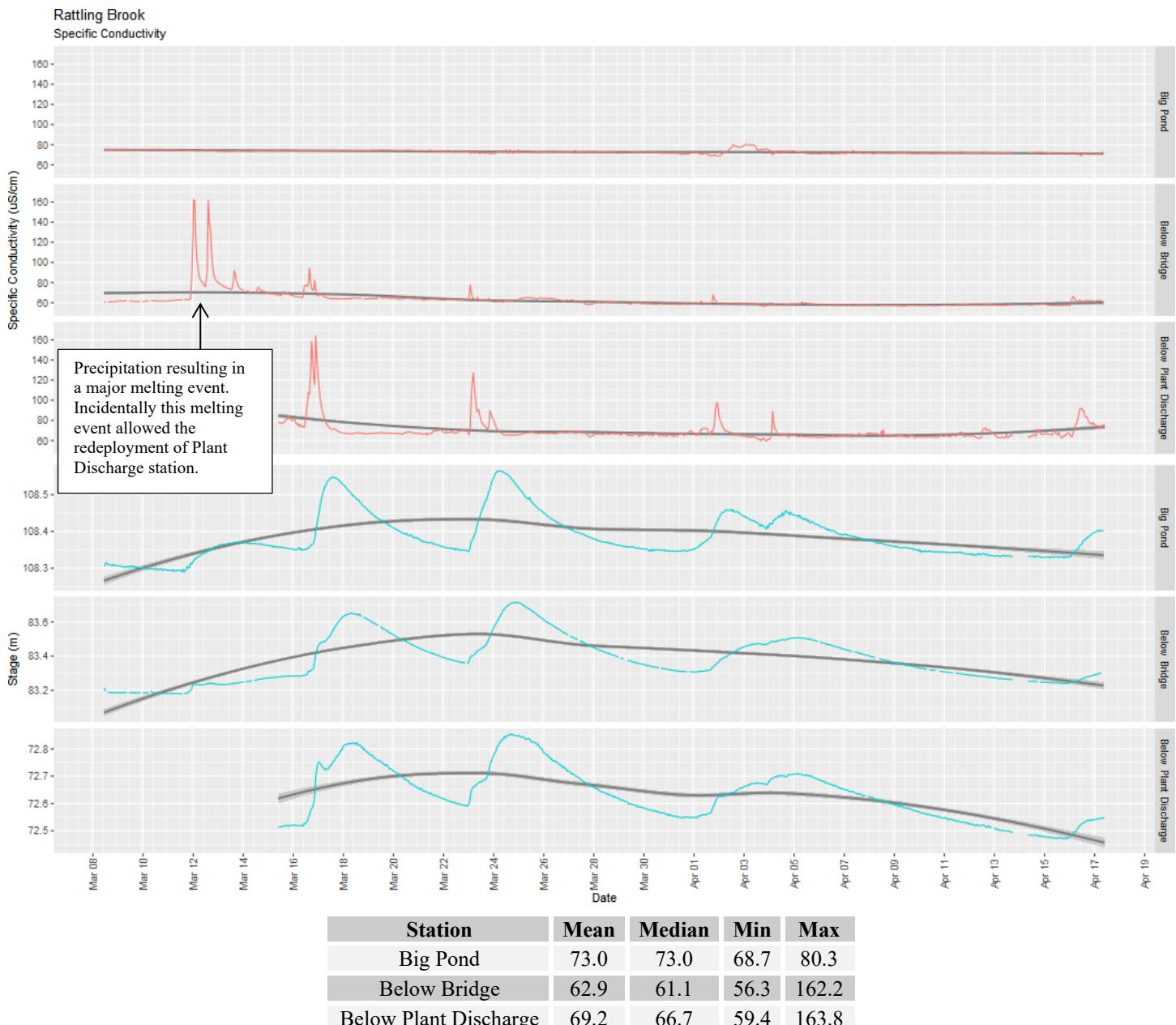


- pH didn't show major deviations at any station during this deployment period, though slow downward trends were observed at Bridge and Plant Discharge stations towards the latter part of the deployment period. pH was consistently below site specific guidelines (SSGs) at Big Pond station and within or just above SSGs at Bridge and Plant Discharge stations.
- A resumption of diurnal pH cycling is especially evident at Bridge station.

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Specific Conductivity

Conductivity relates to the ease of passing an electric charge – or resistance – through a solution. Conductivity is highly influenced by the concentration of dissolved ions in solution: distilled water has zero conductivity (infinite resistance) while salty solutions have high conductivity (low resistance). Specific Conductivity is corrected to 25°C to allow comparison across variable temperatures.

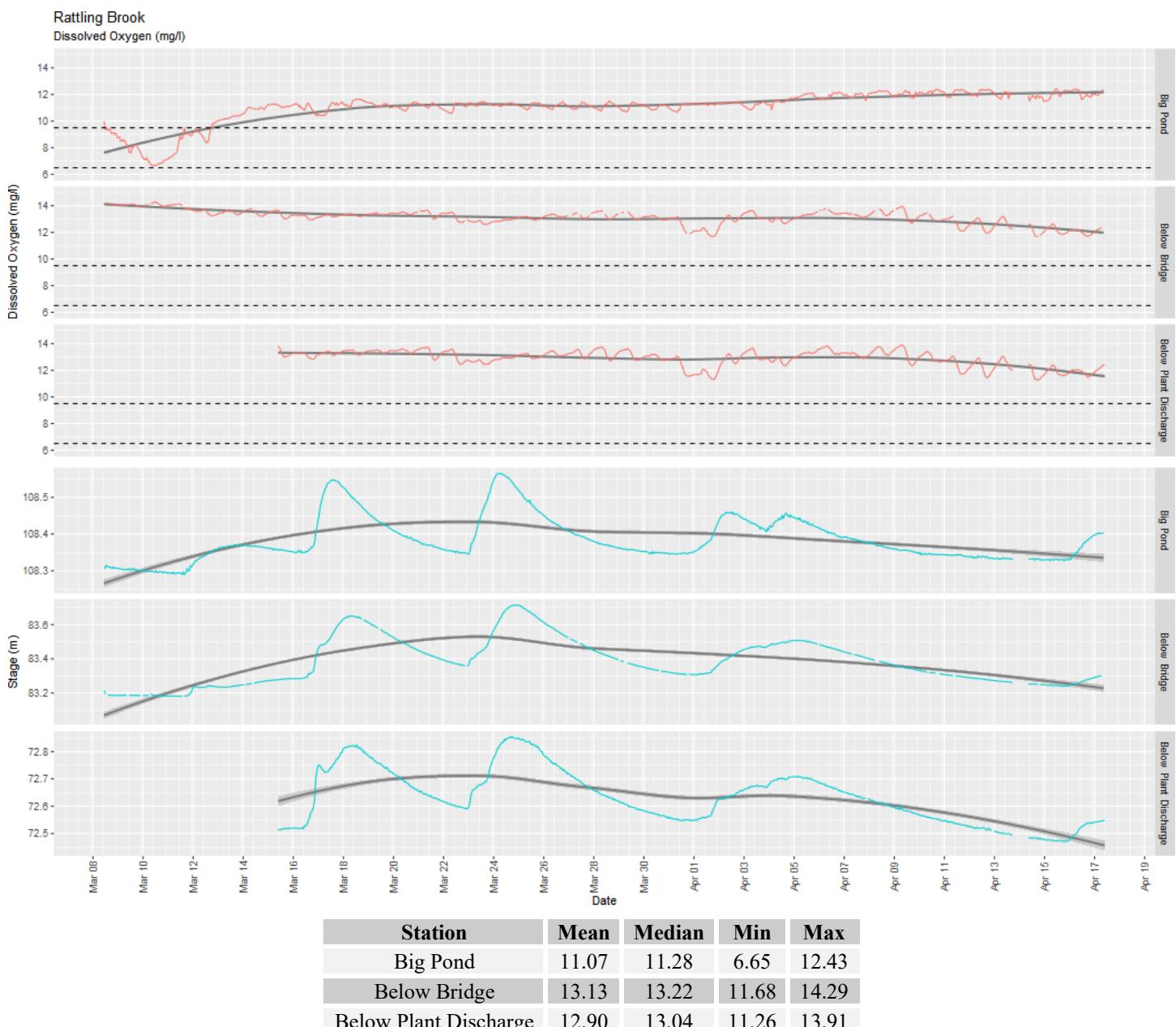


- No major trends were seen in specific conductivity at any station during this deployment period, although there were some peaks observed in relation to environmental events. Additional peaks should be expected during the wet spring season.

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Dissolved Oxygen

Dissolved oxygen is a metabolic requirement of aquatic plants and animals. The concentration of oxygen in water depends on many factors, especially temperature – the saturation of oxygen in water is inversely proportional to water temperature. Oxygen concentrations also tend to be higher in flowing water compared to still, lake environments. Low oxygen concentrations can give an indication of excessive decomposition of organic matter or the presence of oxidizing materials.

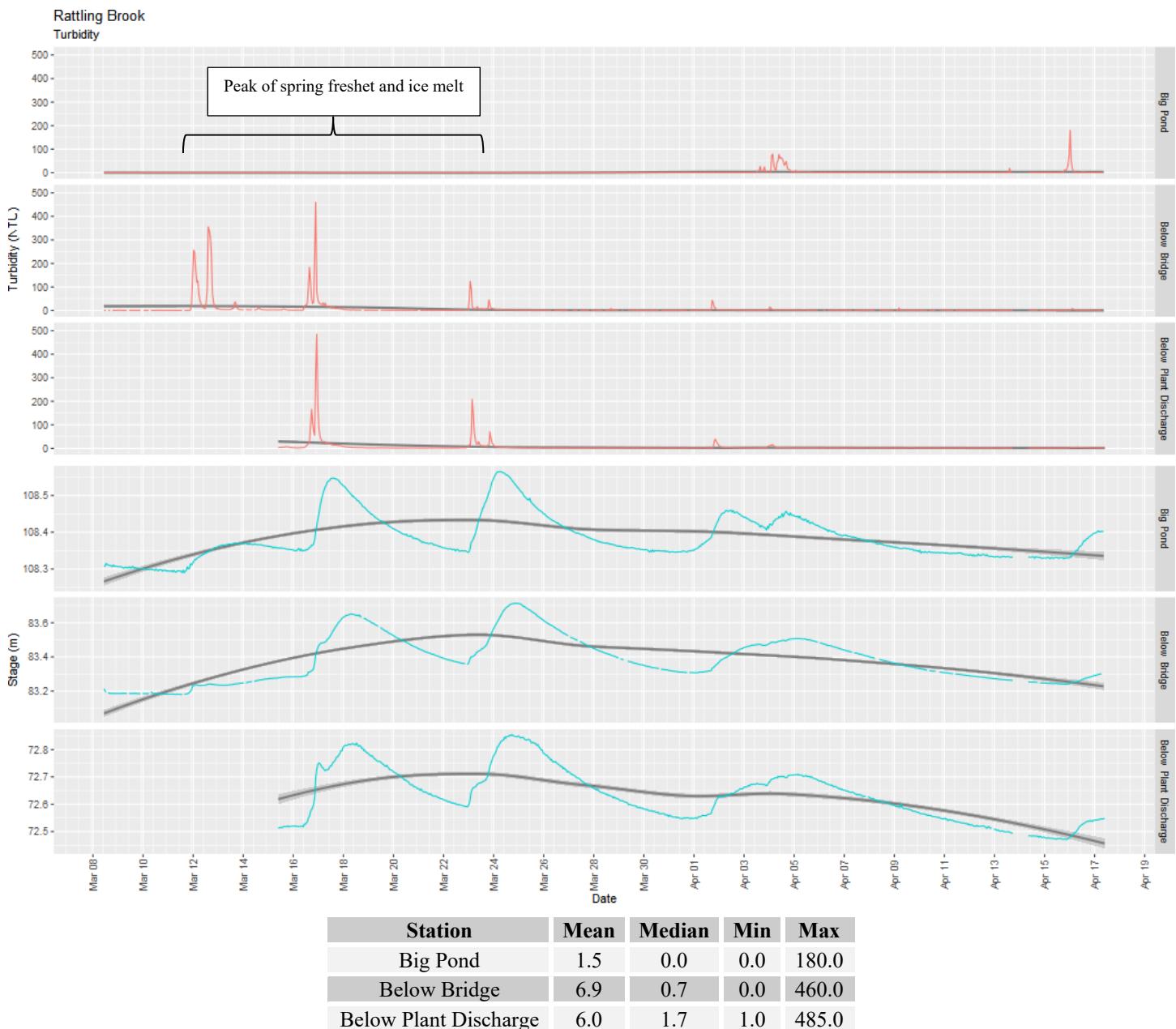


- A period of low- DO was observed at Big Pond station in early March before a rebound brought levels above the CCME guidelines in mid-March. This likely corresponded with the opening of ice at Big Pond allowing the resumption of atmospheric gas exchange.

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Turbidity

Turbidity is typically caused by fine suspended solids such as silt, clay, or organic material. Consistently high levels of turbidity tend to block sunlight penetration into a waterbody, discouraging plant growth. High turbidity can also damage the delicate respiratory organs of aquatic animals and cover spawning areas.



- Turbidity levels were low during this reporting period, especially at Big Pond station. Downstream at Bridge and Plant Discharge stations, turbidity events in excess of 400 NTU were observed, especially during the peak of spring freshet – due to the recession of ice cover and mobilisation of previously-frozen silt and sediments.

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Appendix

